



Development of finely dispersed doped carbon-based materials for the strike point areas of the ITER divertor

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- Introduction
- Why **doped** carbon for ITER?
- Manufacturing route
- Results
- Comparison with candidate materials
- Conclusions and outlook

Carbon based materials (CFC) are present candidates for the strike point areas of the ITER divertor

Advantages of **CFC** materials:

- ❖ Very good thermo-mechanical properties (thermal conductivity at RT > 300 W/mK; high mechanical strength)
 - ⇒ Excellent **power handling capability**
- ❖ No melting under transient power loads and off-normal events (ELMs, disruptions, VDEs)
- ❖ **Low Z** → low radiative power losses in the plasma
- ❖ Broad tokamak operation experience with C-based materials → allows operational flexibility
- ❖ low neutron activation

Main **disadvantages** of C-based materials for fusion application (graphite and/or CFC) :

- ❖ **CFC, graphite**: chemical reactivity with H → erosion, tritium co-deposition → the divertor is **not** a permanent component
- ❖ **Graphite**: thermal conductivity not high enough (>300 W/mK at RT needed)
- ❖ **CFC**: anisotropy
 - unequal, partly severe erosion by **brittle destruction**
 - joining process to heat sink more difficult
- ❖ **CFC**: high **cost**
- ❖ **CFC, graphite**: neutron damage > 0.5 dpa → loss of thermal conductivity, dimensional instability

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Possible way for **improving carbon materials for fusion application**: addition of **metallic or ceramic dopants**.

Effects of **dopant addition** (several at.%):

❖ **Self-passivation** against **chemical erosion by hydrogen**.

Passivation occurs due to

- Influence of dopants on **thermal activated** hydrocarbon release
- **Dopant enrichment** at the surface due to preferential erosion of C ▶

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Both effects are the more effective the smaller the dopants particle size

- ❖ Improvement of **thermal conductivity** → dopants showing **catalytic effect** on the graphitization
- ❖ Carbide addition → reinforcement effect ⇒ **improved strength**

In all cases improvement the more effective the smaller the particle size of the dopants  **nanoscaled dopant distribution**



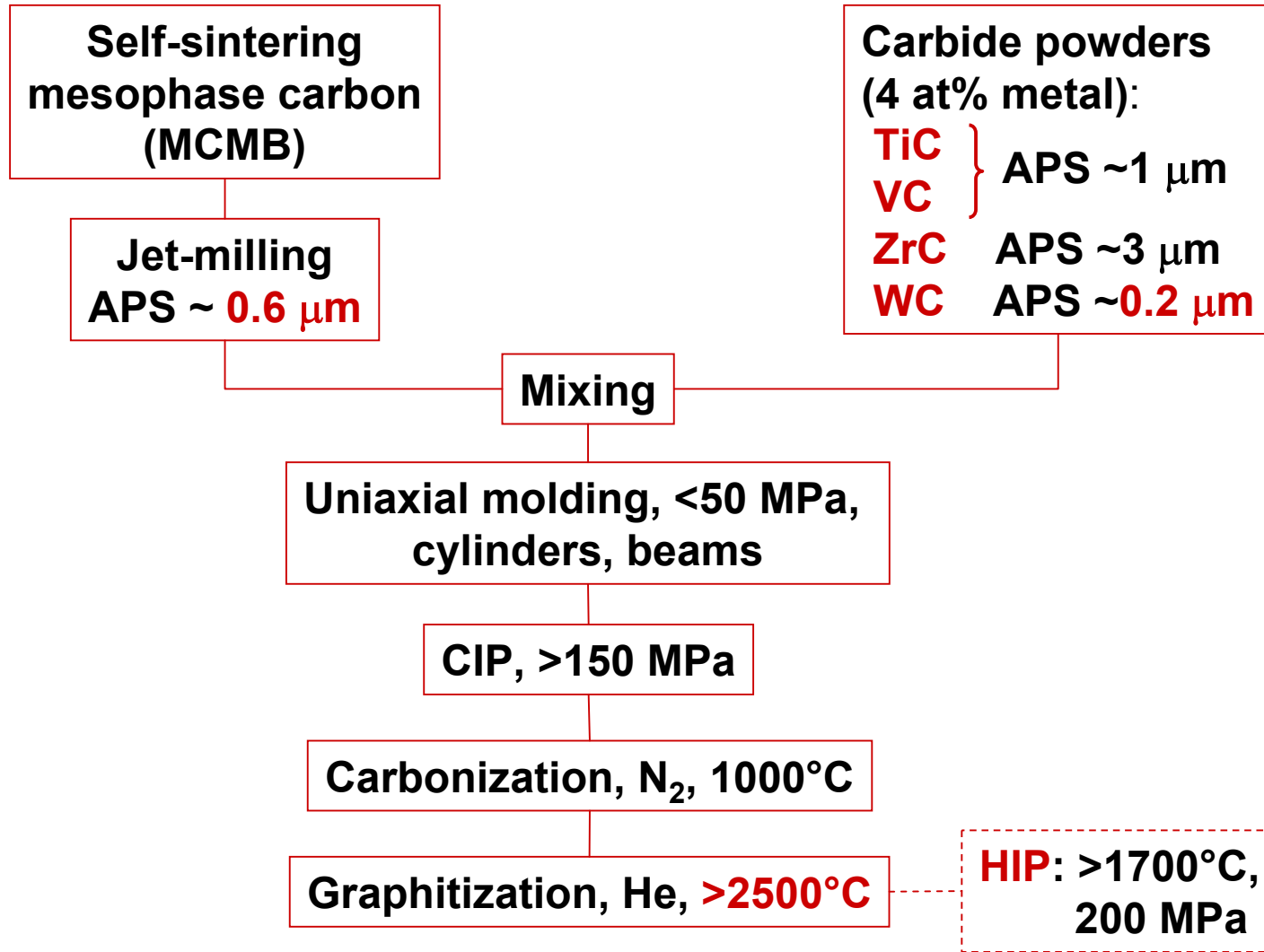
Aim of this work: development and **optimization** of fine-grained isotropic **graphite** for **plasma facing application** by **doping** with finely dispersed **carbides**

 **reduction of chemical erosion** by H while **improving thermo-mechanical properties** (high thermal conductivity, high thermal-shock resistance); **reduced costs** compared to CFC materials.

Keys for development:

- ❖ Selection of appropriate **dopants** and raw materials
- ❖ Use of starting powders with **very small particle size**
- ❖ Rigorous **control of all manufacturing steps**

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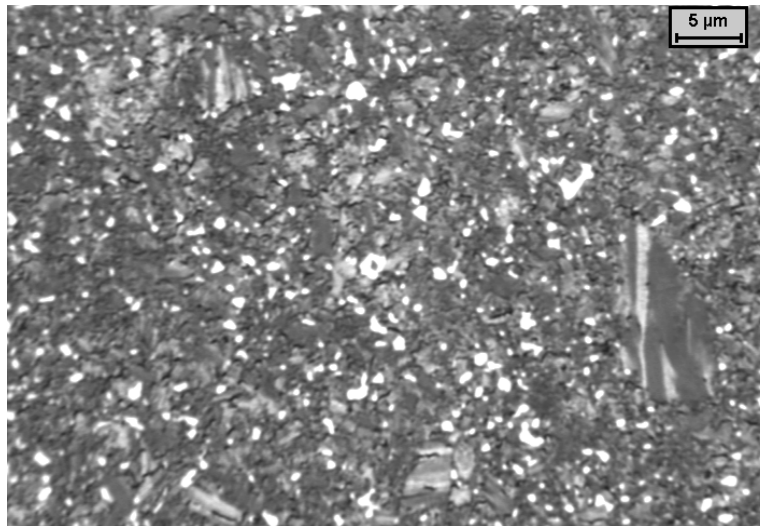
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Parameters investigated after different graphitization cycles:

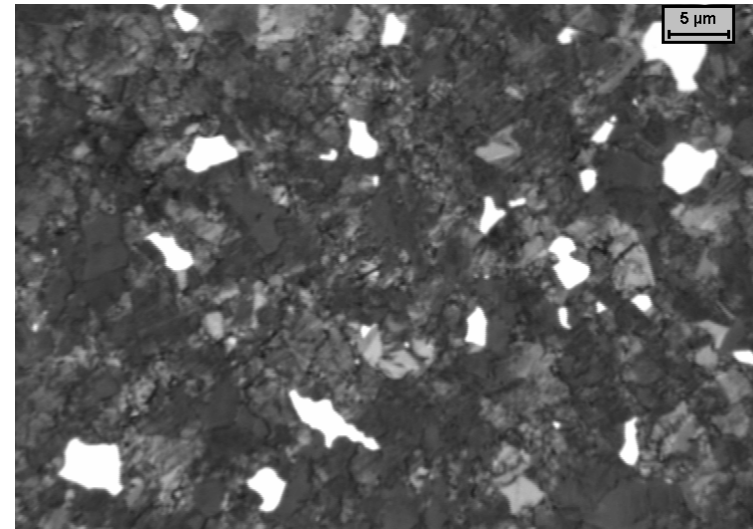
- ❖ **Carbide distribution**
- ❖ **L_c , thermal conductivity**
- ❖ **Porosity** (total, open)
- ❖ **Influence** of **dwel time** during graphitization, and of a **HIP** step subsequent to graphitization
- ❖ **Mechanical properties** (flexural strength, Young's modulus, strain-to-failure), **fracture surfaces** → responsible defect.
- ❖ Optimum **dopant concentration** (only for **TiC**)

Carbide distribution:

- Homogeneous distribution after carbonization (1000°C)
- After **graphitization** → strong influence of cycle parameters
(T_{graph} , **dwel time**): longer dwell time ⇒
 - ❖ Coarsening



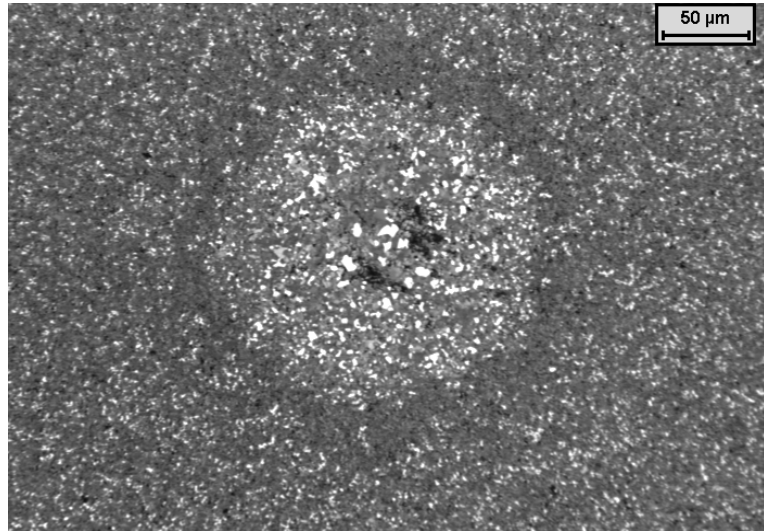
1 h dwell



2 h dwell

TiC-doped sample after graphitization at 2650°C

- ❖ Local agglomeration, carbide depleted zones → detrimental for **mechanical properties**

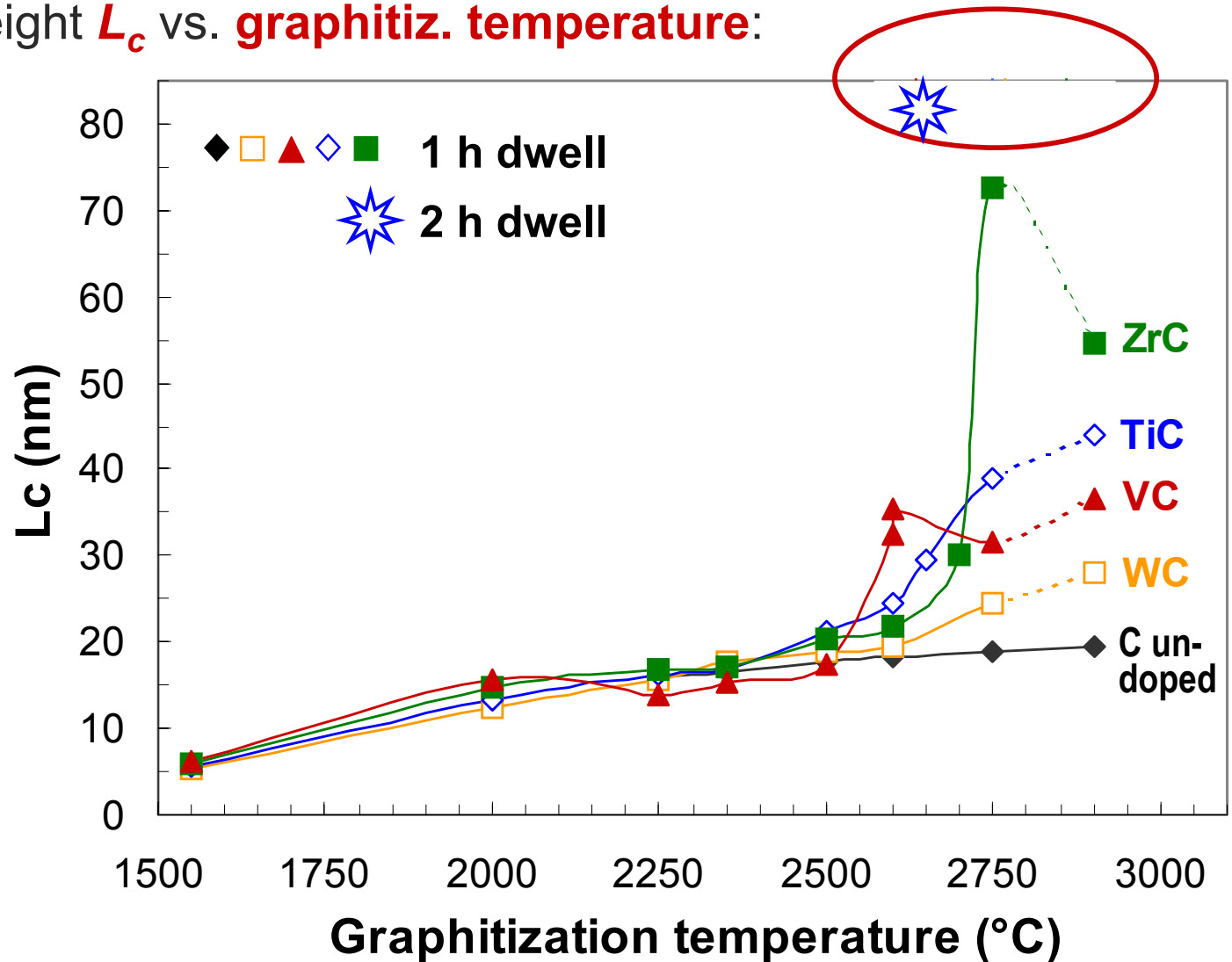


VC-doped sample after graphitization
at 2600°C

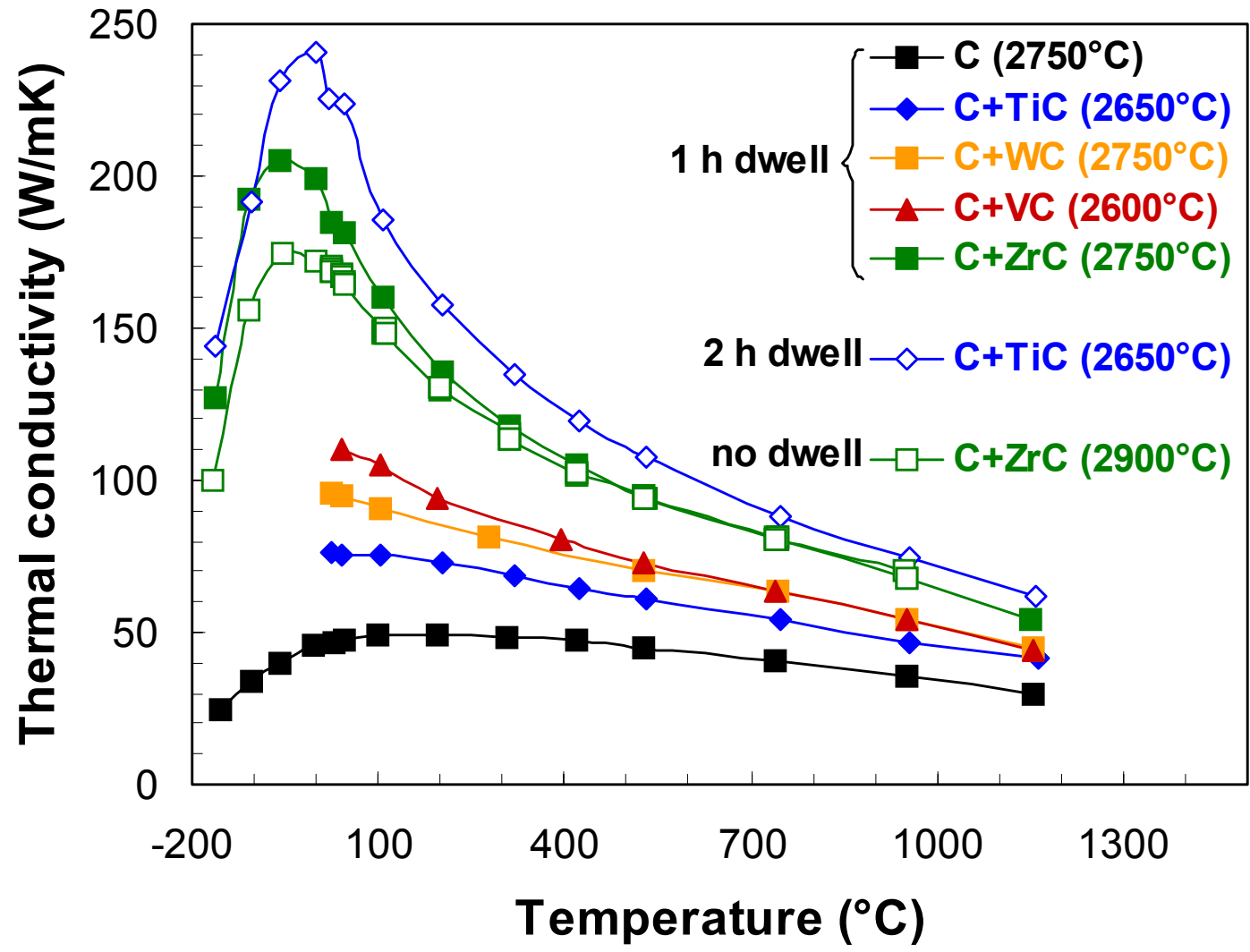
⇒ Necessary to find **compromise** between **high T_{graph}** and **long dwell time**

Crystallite height L_c vs. **graphitiz. temperature**:

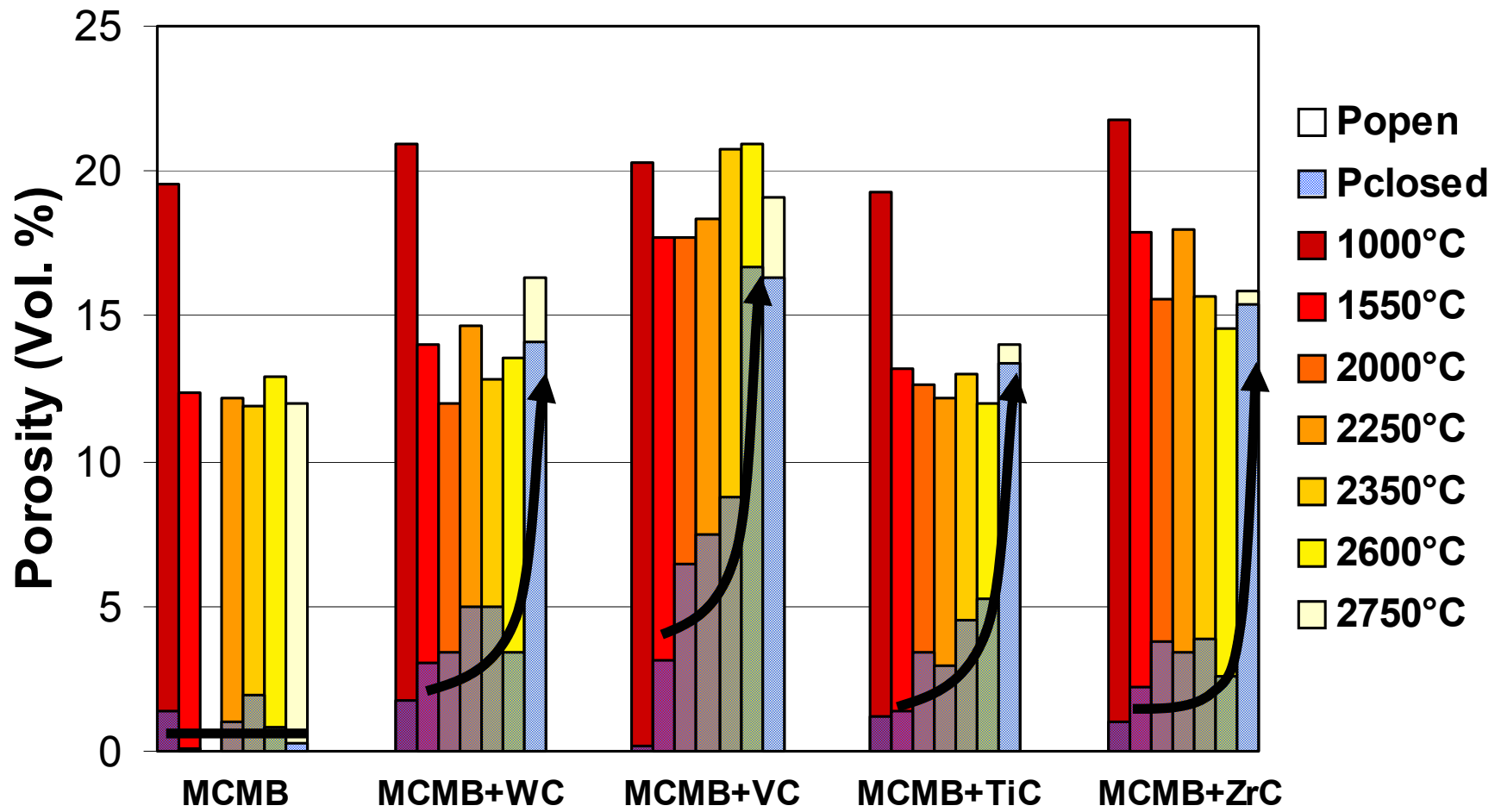
L_c : X-ray diffraction (Scherrer)



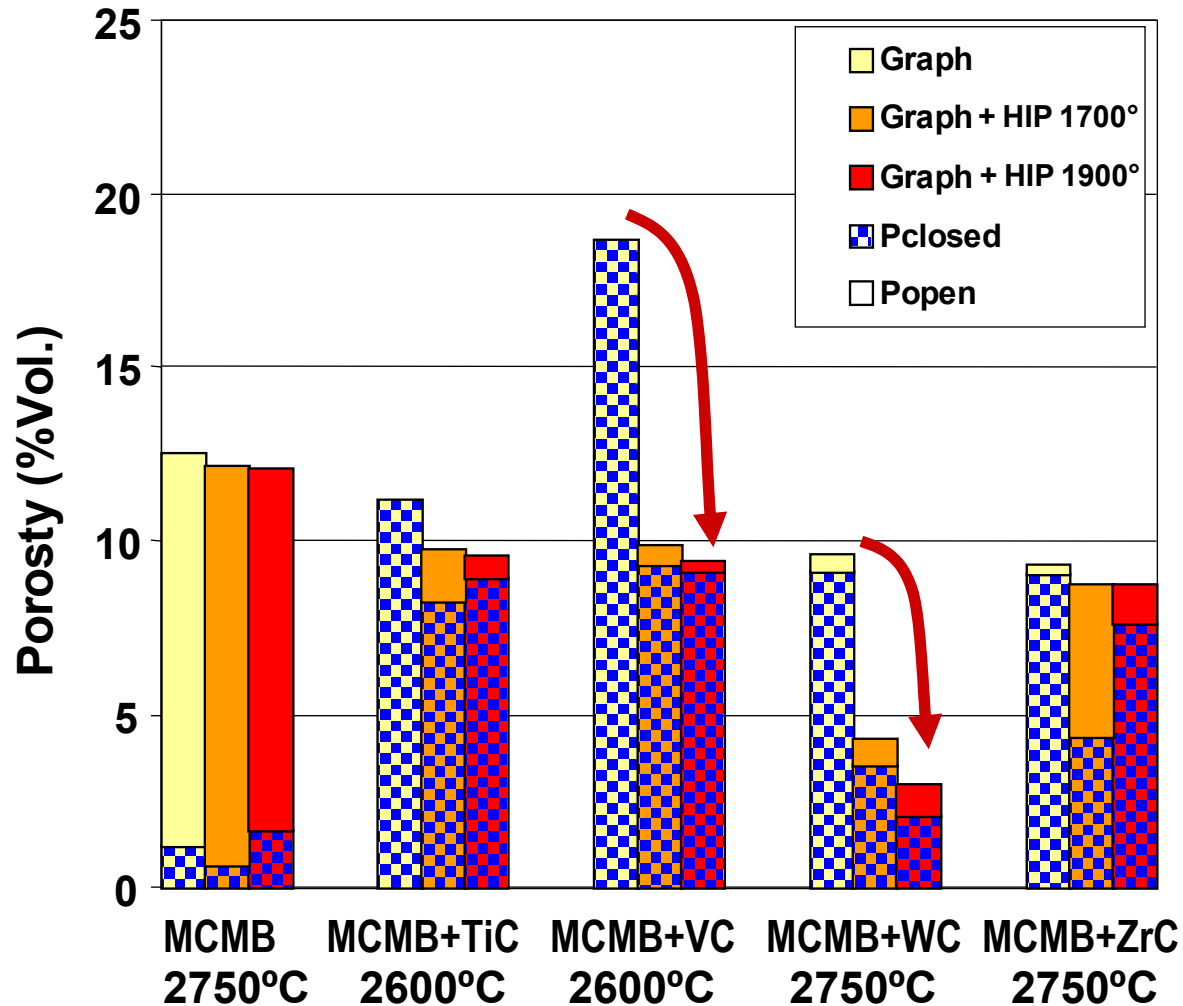
Thermal conductivity:



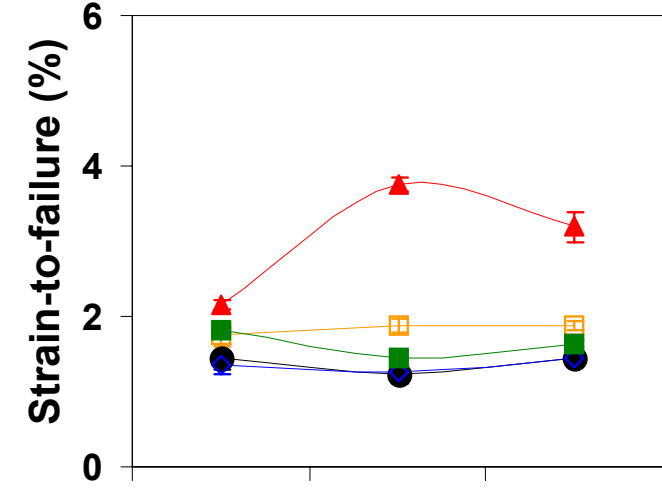
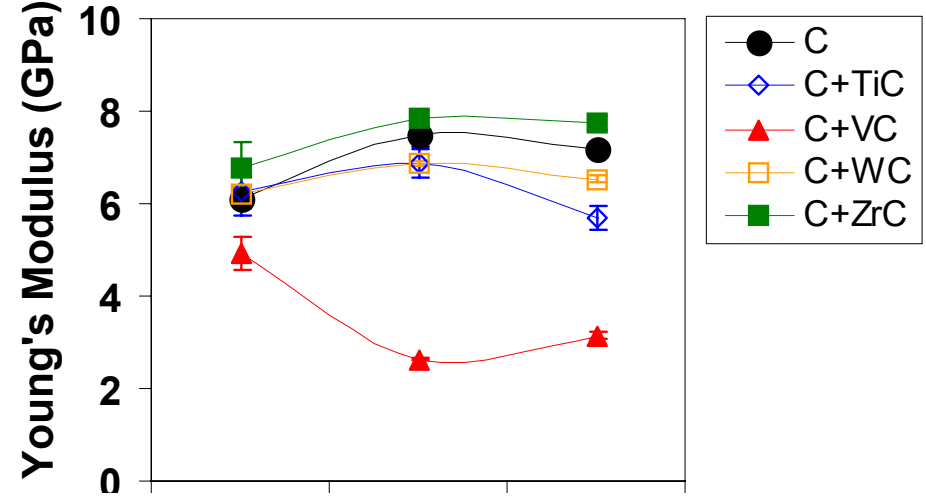
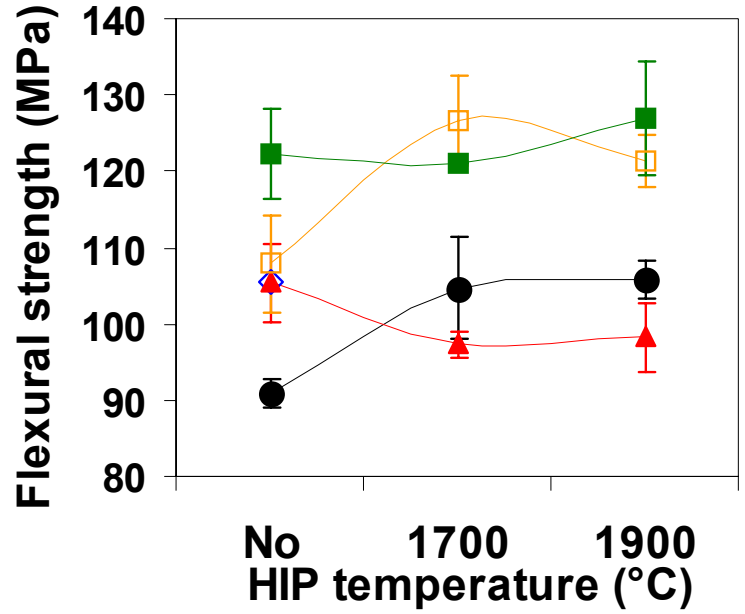
Total and open porosity after graphitization at different temp.:



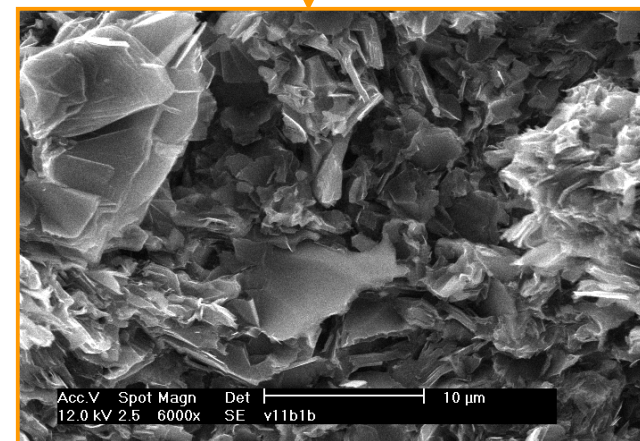
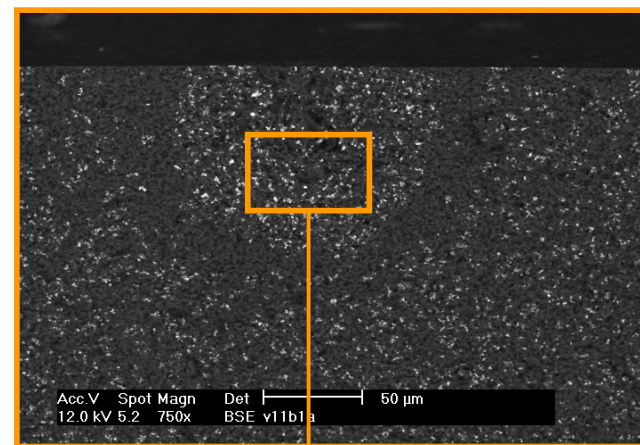
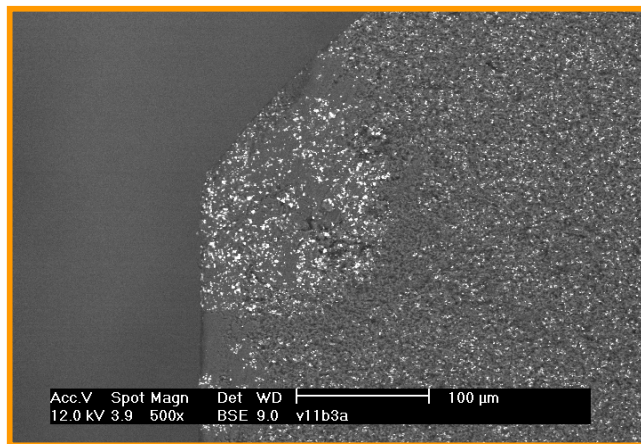
Decrease of porosity by HIP subsequent to graphitization:



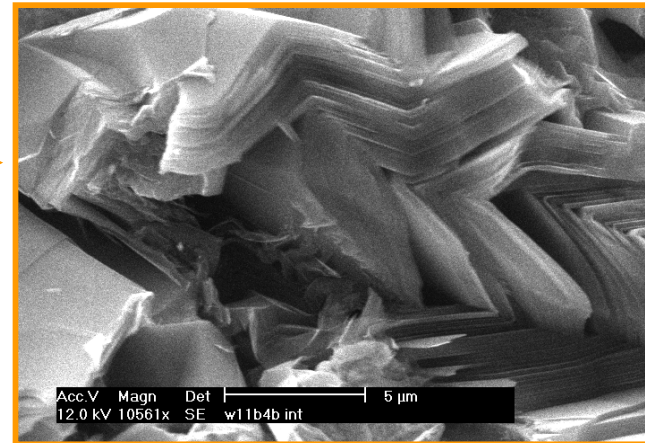
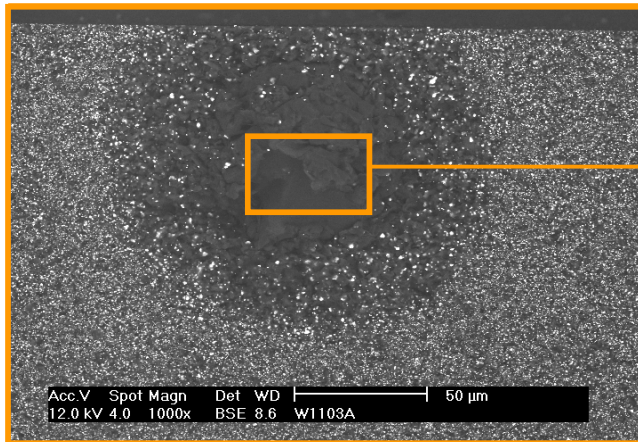
Mechanical properties; influence of HIP



Fracture surfaces: VC-doped material

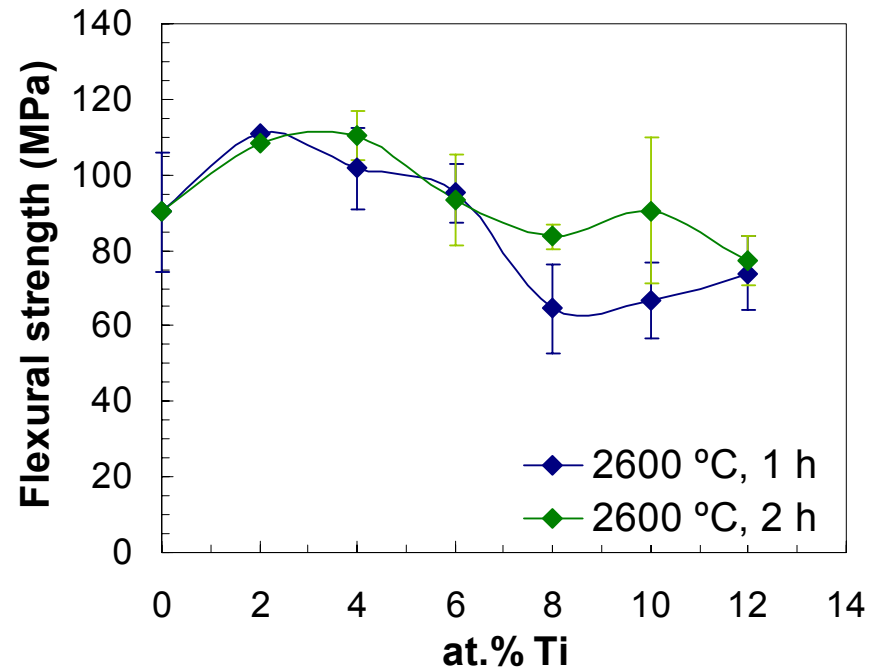
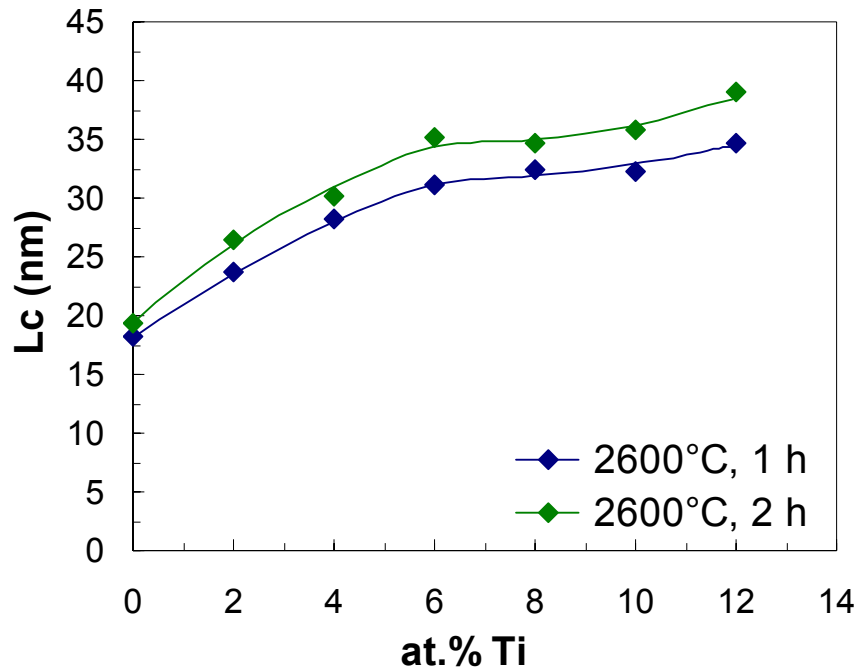


Fracture surfaces: WC-doped material



Determination of **optimum dopant concentration**

→ variation of **Ti** content between 2 and 12 at% Ti



➔ Optimum Ti concentration (for TiC APS $\sim 1 \mu\text{m}$): **$\sim 4 \text{ at.\% Ti}$**

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Comparison with present candidate materials:

❖ European reference CFCs: **NB31** and **Concept 2**

		Density (g/cm ³)	Thermal conduct. (W/mK)	Strength (MPa)	Young's modulus (GPa)
NB31	Pitch	1.90	323	σ_t 130	107
	PAN			30	15
	Needling			19	12
Concept 2	Pitch	1.86	360	σ_t 106	
	PAN			57	
	Needling			13	
Doped graphites		1.92 (undoped)	50 - 230	σ_f 100 - 125	3 - 8

Advantages: low chemical erosion + good thermo-mechanical properties + isotropy + low cost; further improvement possible

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- **Doping** → efficient way for improving isotropic graphite such as to achieve specifications of candidate C-based materials for ITER
- **Longer dwell time** during graphitization → ↑ L_c → ↑ **thermal conductivity**. But **carbide coarsening and agglomeration**, depending on temperature → ↓ **flexural strength** ⇒ balance between dwell time and graphitization temperature to obtain optimum properties.
- **HIP** subsequent to graphitization → ↓ **porosity** → improved mechanical properties: ↑ **flexural strength**, ↓ **Young's modulus**, ↑ **strain-to-failure** ⇒ ↑ **thermal shock resistance**.
- Further improvement expected with **nanoscaled dopant distribution** ➡ **ExtreMat IP** → extension to other applications



END

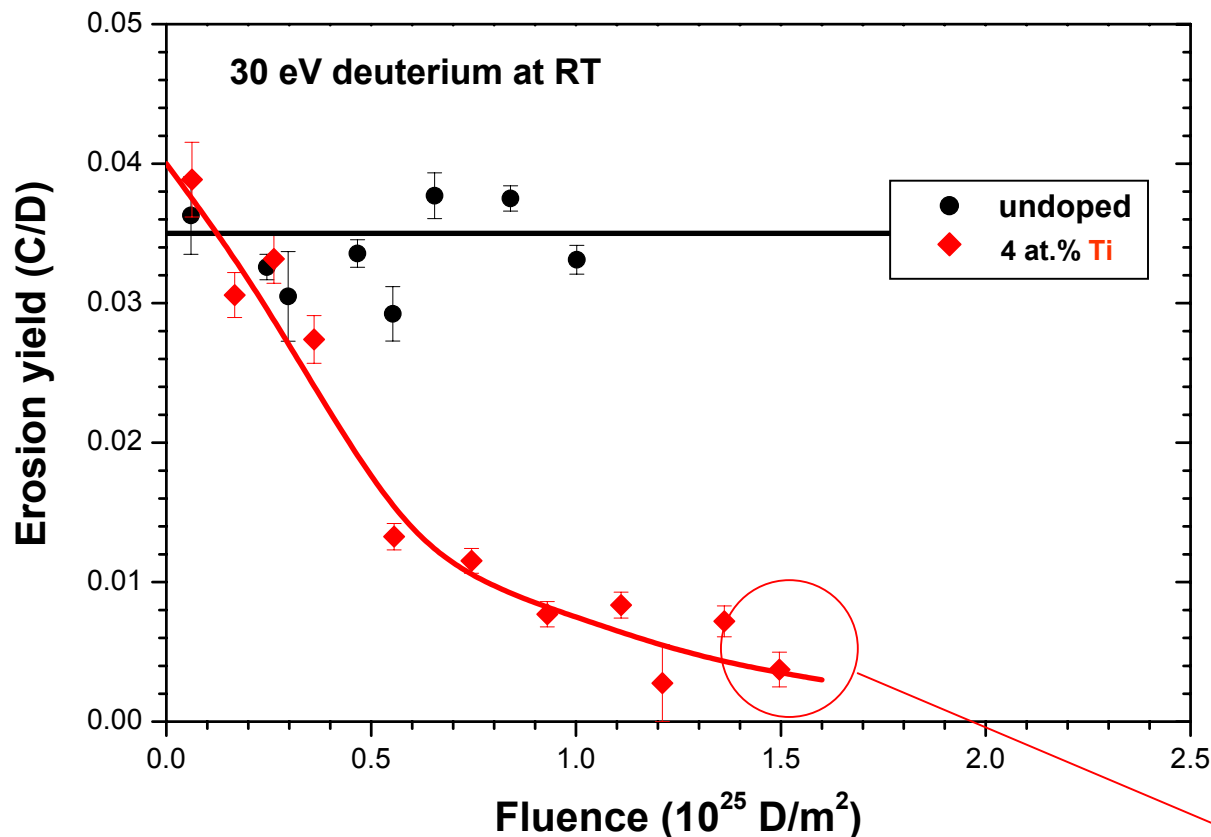
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Fluence dependence of chemical erosion yield

(E. de Juan Pardo et al., Physica Scripta T111, 2004)



columnar surface topography, where carbide grains protect underlying graphite from further erosion

