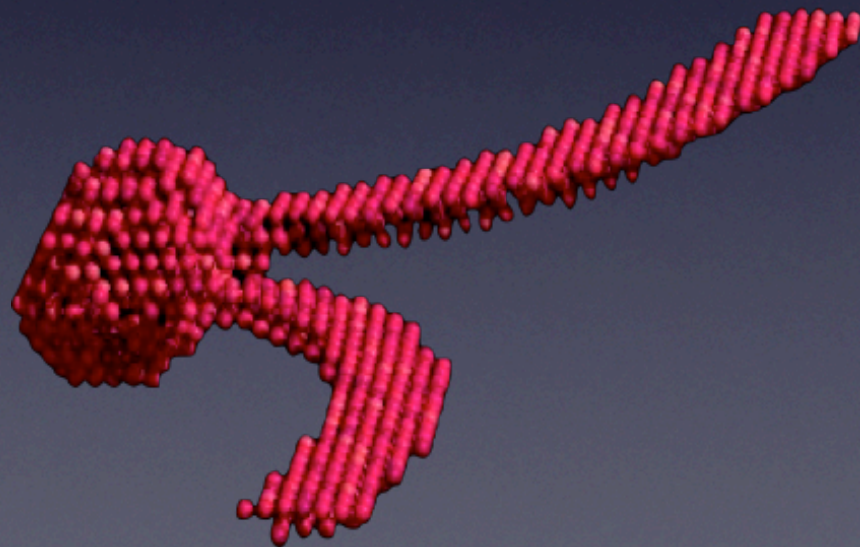


# Molecular dynamics simulation of Fe plasticity in the presence of multiple defects

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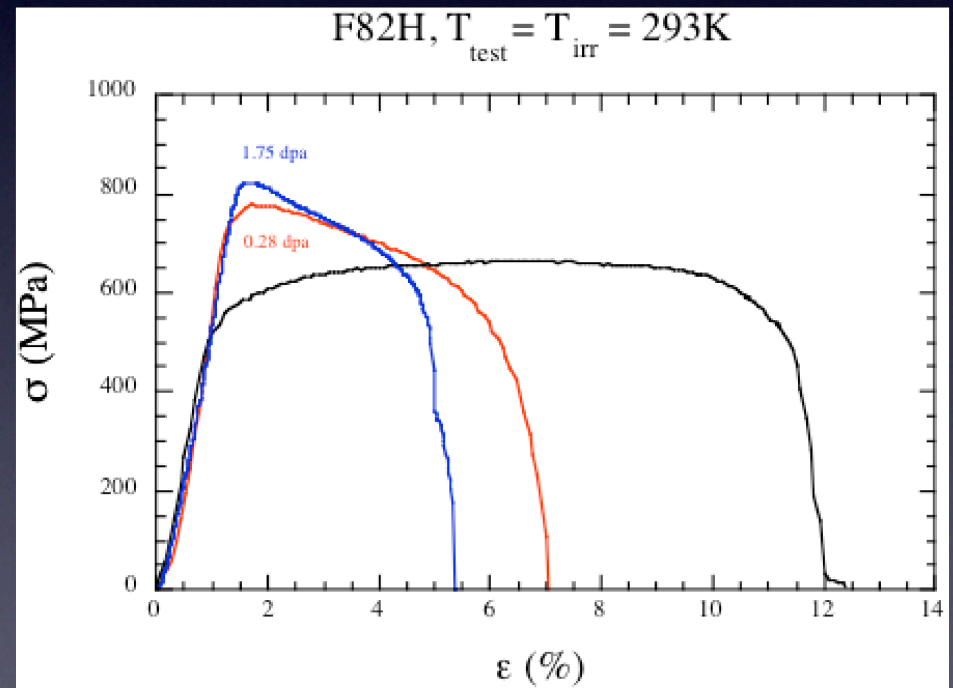


# Overview

- Introduction
- Molecular dynamics simulation technique
- Simulation conditions
- Edge dislocation–defect interaction
  - Interaction with void
  - Interaction with He bubble
  - Interaction with Cr precipitate
- Summary

# Introduction

- Irradiation-induced defects as voids, He bubbles and Cr precipitate → substantial **hardening** & **loss of ductility** below 400°C.
- Ferritic base steels → prime candidates for the plasma facing materials & the structural components of the future fusion reactors.
- Multiscale modeling → major tool in description of radiation defects in materials.
- Aim: to obtain a relationship between the irradiation defects and mechanical behavior by MD simulation & to get a physical concept of irradiation effects on Fe.



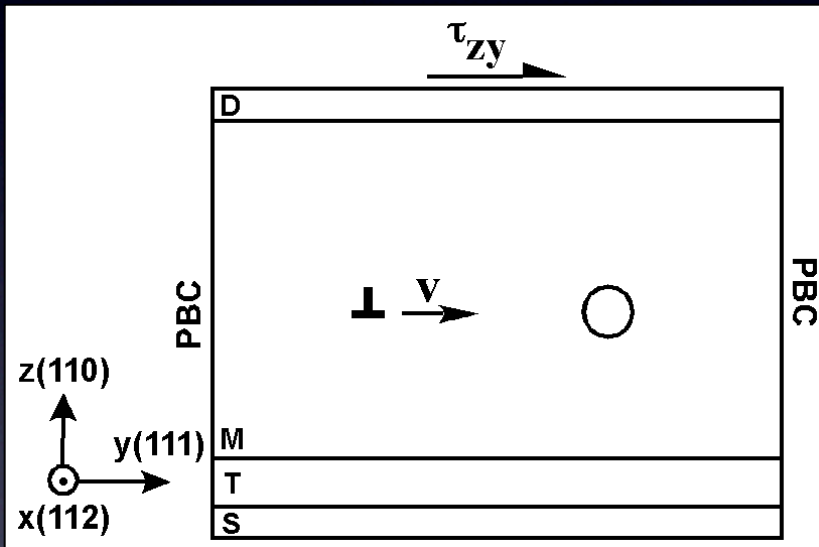
F82H, a ferritic/martensitic steel,  
P. Spätig, R. Schäublin, S. Gyger and M. Victoria  
Journal of Nuclear Materials, 258-263 (1998) 1345-1349

# Molecular Dynamics simulation

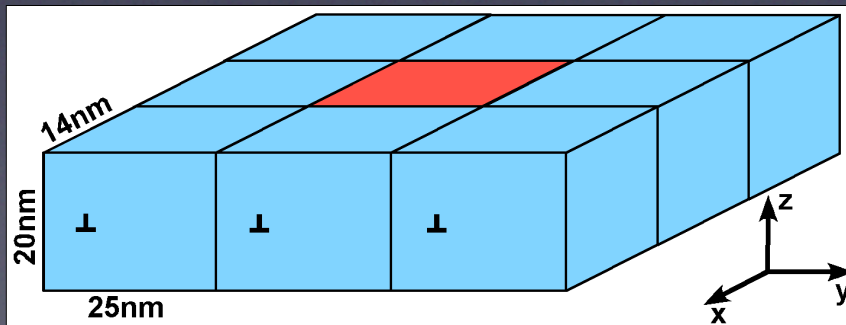
- MD simulation: solving the Newton equation of motion for particles of a system using empirical interatomic potentials, which gives trajectory of atom displacement
- Simulation steps :
  - Sample creation → describes the atomic configuration of dislocation (Disloc)
  - Deformation → allows the dislocation to move, using the embedded atom method (Moldy)
- Deformation mode: shear due to an applied strain rate in the upper region of sample.
- Material: bcc-Fe, single crystal

# MD simulation sample

- A box including an edge dislocation in [112] direction & a defect on the dislocation slip plane, (110)



- The box is built up in several regions:
  - The mobile region: atoms follow Newton equation
  - Upper region: atoms to control the deformation of sample and are forming a free surface
  - Bottom region 1: the thermal bath
  - Bottom region 2: static atoms to anchor the specimen



- Periodic boundary condition applied along dislocation line direction & slip direction

# Simulation conditions

- The many-body interatomic potentials:
  - Fe-Fe → Ackland 1997, Mendeleev 2003, Dudarev-Derlet 2005
  - Fe-He → Wilson-Johnson 1972
  - He-He → Beck 1968
  - Fe-Cr & Cr-Cr → Olsson 2005
- MD simulation parameters:
  - Defect type → void, He bubble, Cr precipitate
  - Temperature → 10, 100, 200, 300, 500, 700 (K)
  - Defect size → 1, 2, 3, 4, 5 (nm)
  - He density → 0, 1, 2, 3, 4, 5 (He/v)

Size of box	$(14-17) \times 25 \times 20 \text{ nm}^3$
No. of Fe atoms in the box	566586
Dislocation speed	$60 \text{ m} \cdot \text{s}^{-1}$
Time step	1 fs
Annealing before straining	5 ps
Total simulation time	480 ps

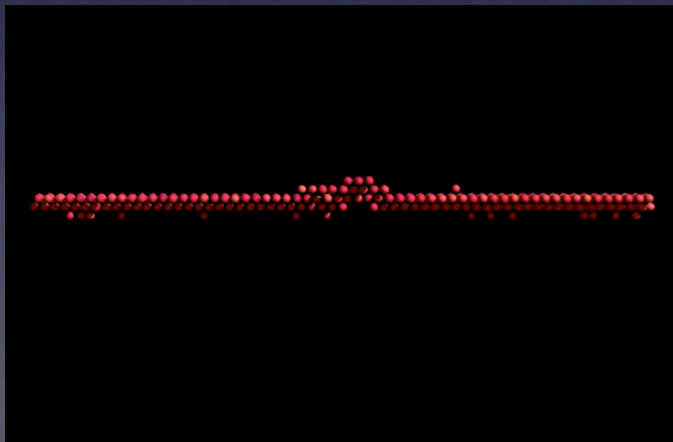
# Dislocation – void Interaction

- Each passage of dislocation shears the void by one Burgers vector.

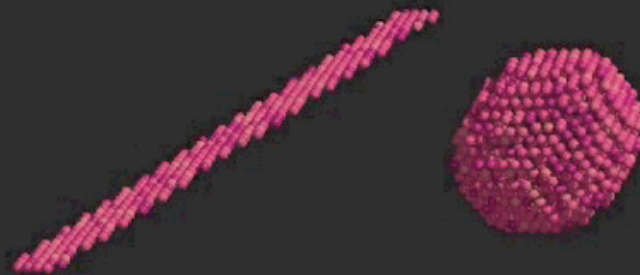
Main interaction mechanism

↓  
Climb & Cross-slip

↓  
Jog formation



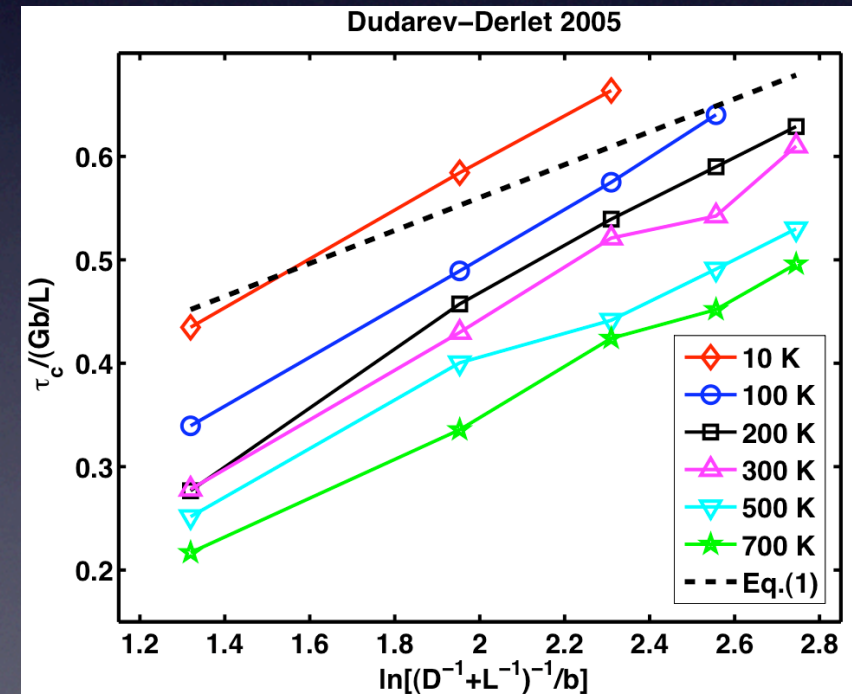
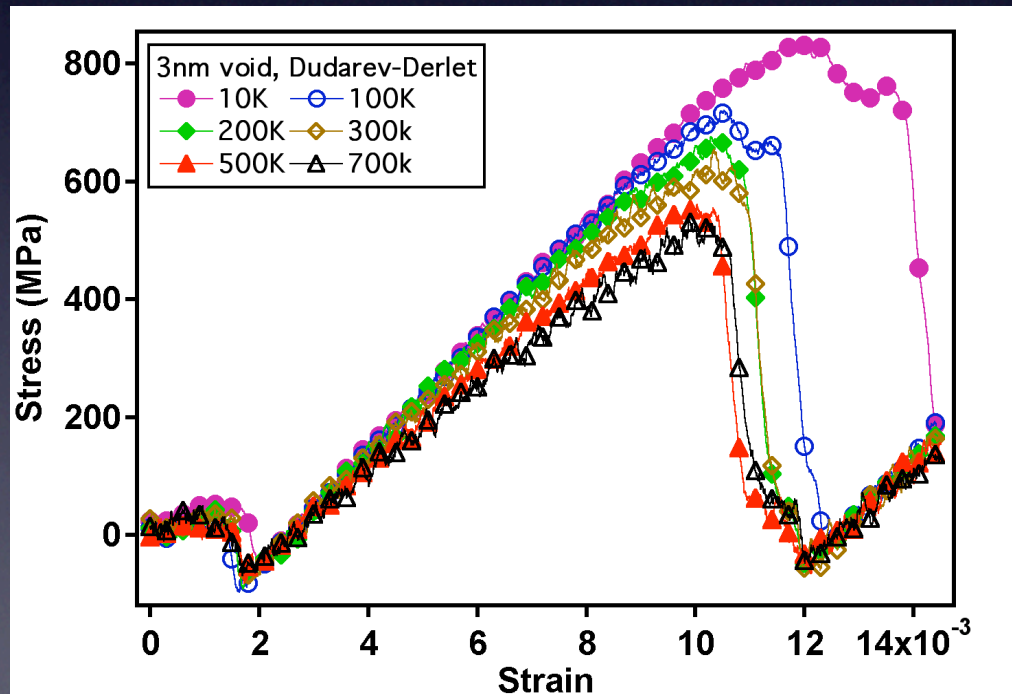
3 nm void 10 K (© 2009 SM Hafez H, CRPP-EPFL)



# Temperature and size effects

- Temperature  $\uparrow \Rightarrow$  release stress and strain  $\downarrow$
- Void size  $\uparrow \Rightarrow$  release stress  $\uparrow$
- Results are compared to elasticity of continuum:

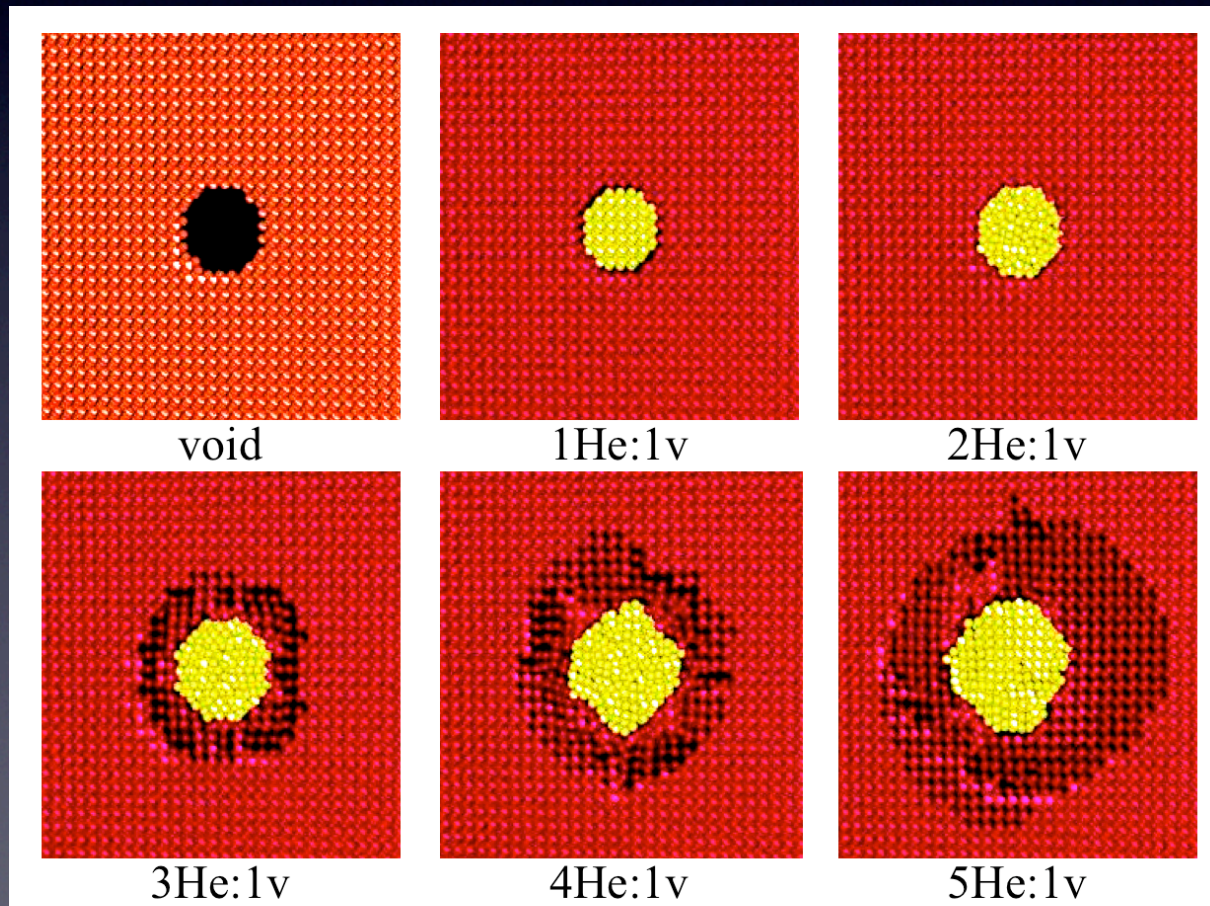
$$\tau_c = \frac{Gb}{2\pi L} \left[ \ln \frac{(D^{-1} + L^{-1})^{-1}}{b} + B \right]$$





# He bubble in Fe

- According to *ab initio* calculations He ratio of 1–5He:1v is reachable
- Initial cavity size: 2 nm
- Atomic structure of He atoms and surrounding Fe atoms after relaxation



S. M. Hafez Haghigat, G. Lucas,  
and R. Schaeublin, *Europhysics  
Letters* 85 (2009) 60008.

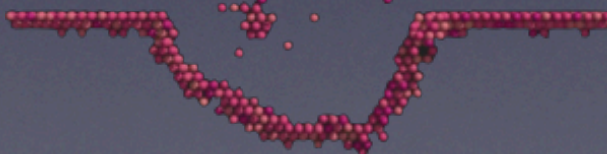
# Dislocation – He bubble Interaction

Jog formation on edge dislocation due to interaction with He bubble

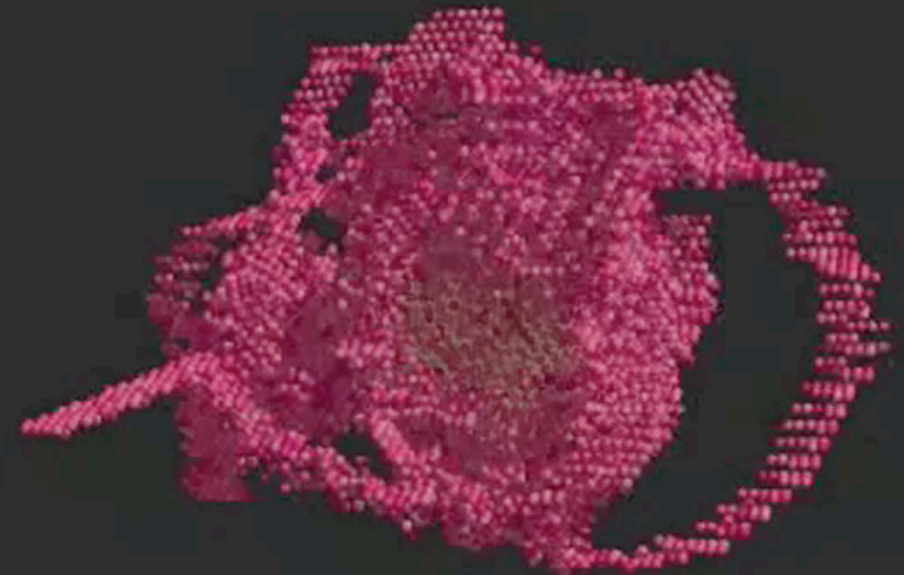
1 He/v



5 He/v



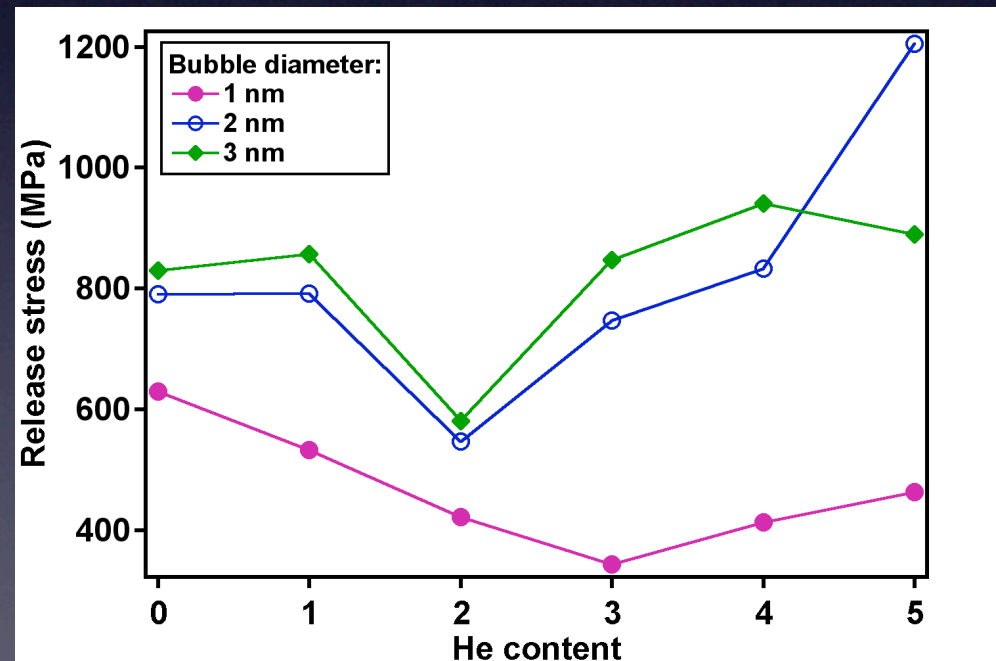
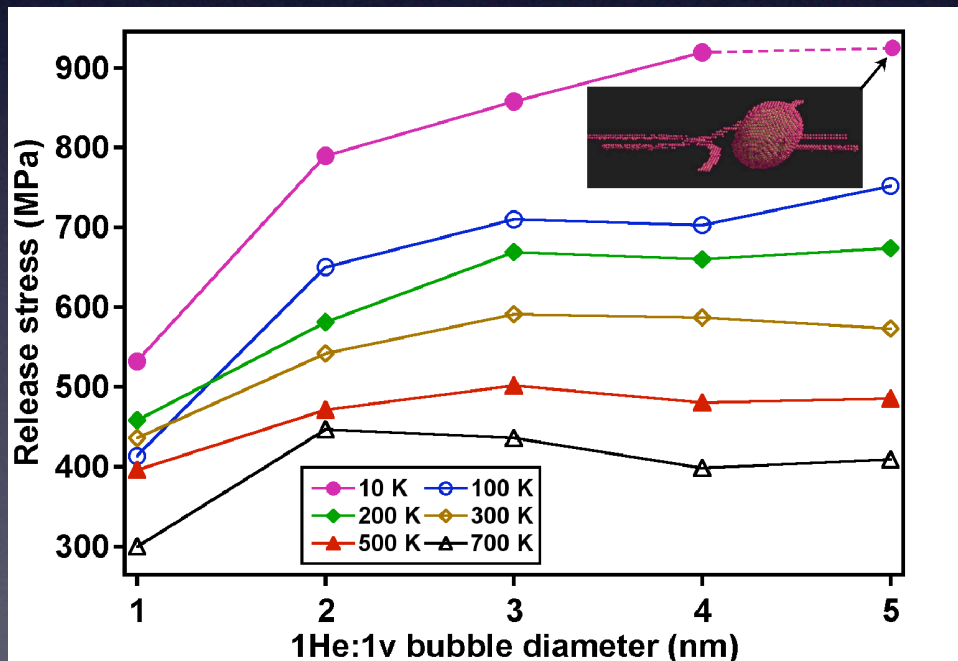
4 nm 5He:1v bubble 10 K (© 2009 SM Hafez H, CRPP-EPFL)



- He content  $\uparrow \Rightarrow$  loop punching (emission of interstitials or dislocation segments from bubble)

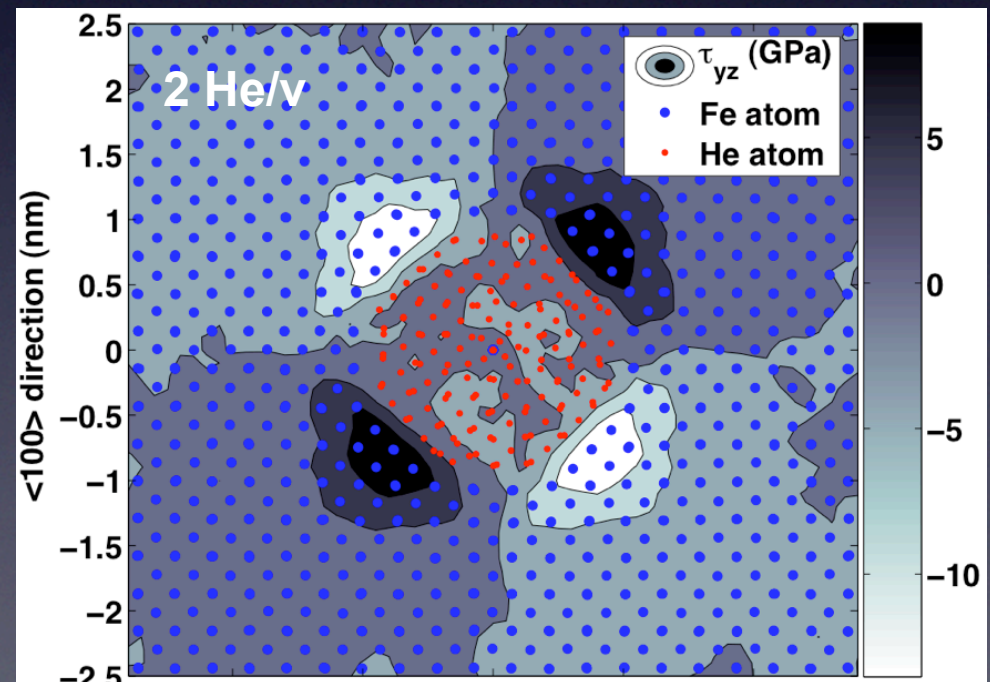
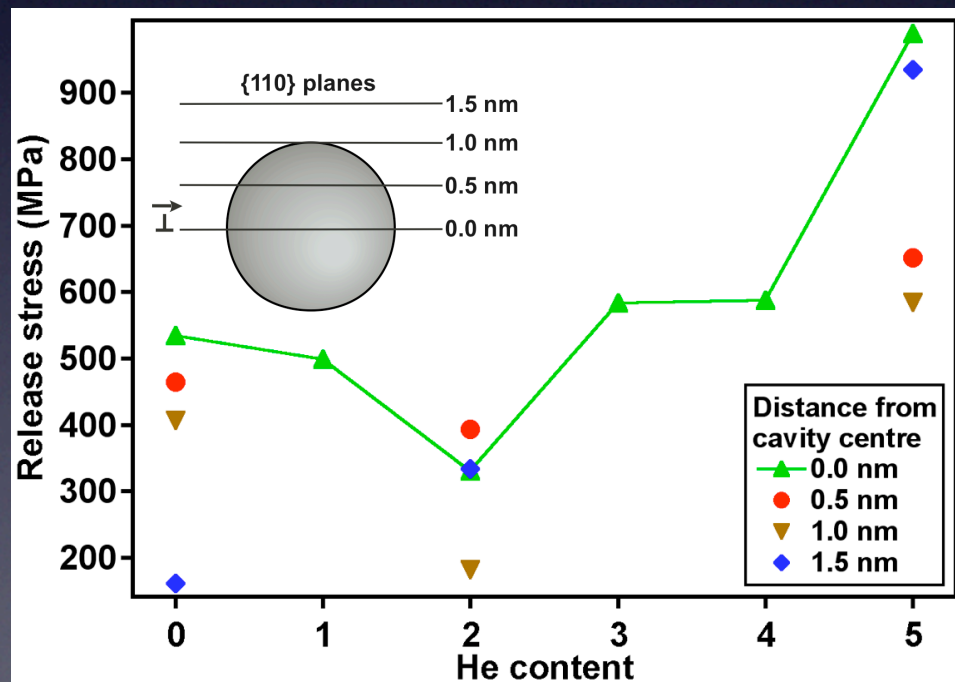
# Temperature and size effects

- Bubble size  $\uparrow \Rightarrow$  release stress  $\uparrow$ , strengthening rate  $\downarrow$
- Drop at moderate He contents (densities) due to surrounding stress field
- High He contents  $\Rightarrow$  the largest obstacle strength due to loop punching



# Dislocation – bubble geometry effect

- Low He content : dislocation height  $\uparrow \Rightarrow$  monotonous softening  $\Rightarrow$  shearing area effect
- Moderate He content : dislocation height  $\uparrow \Rightarrow$  non-monotonous variation  $\Rightarrow$  stress field effect
- High He content : dislocation height  $\uparrow \Rightarrow$  non-monotonous variation  $\Rightarrow$  loop punching effect



# Dislocation – Cr precipitate Interaction

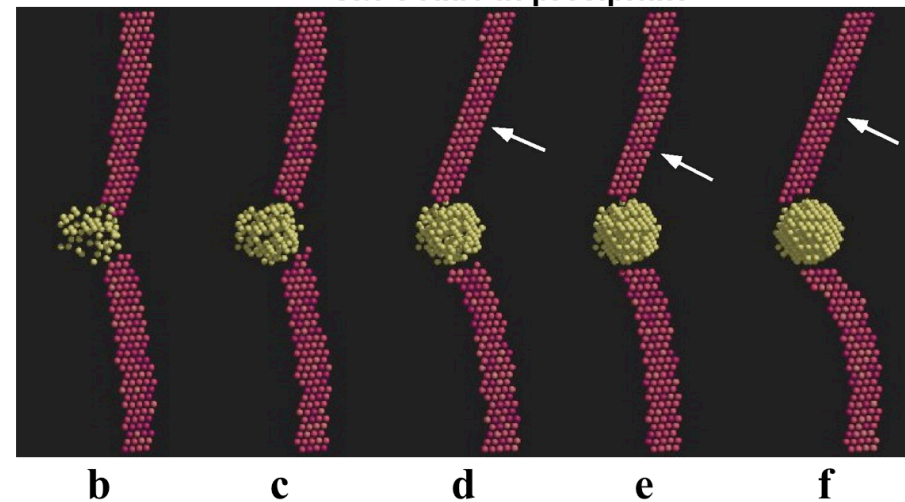
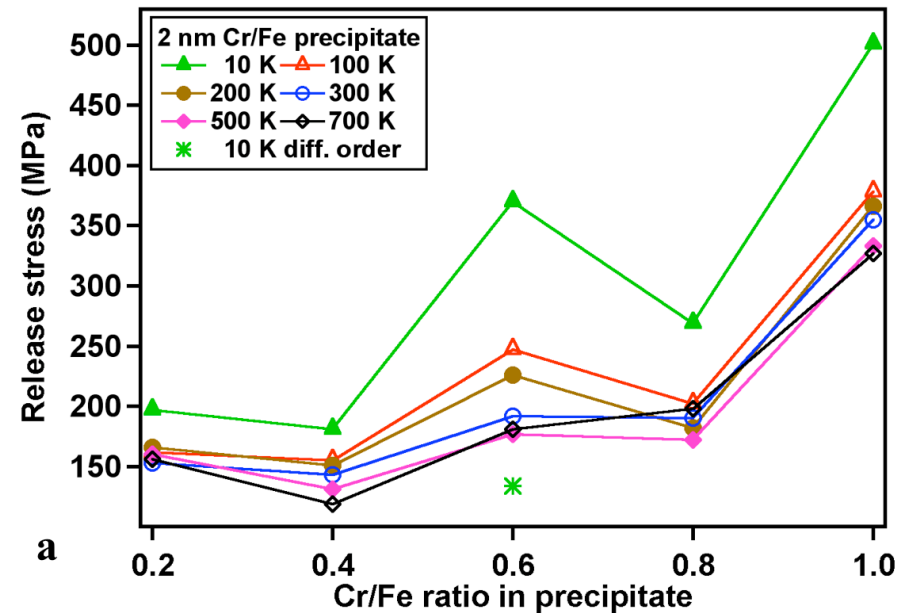
- Cr precipitate is coherent with surrounding Fe.
- No jog formation was observed after dislocation–precipitate interaction
- Shearing of precipitate by one Burgers vector similar to void.

2 nm 0.4Cr/Fe precipitate 10 K (© 2009 SM Hafez H, CRPP-EPFL)



# Effect of Cr/Fe ratio

- With increasing Cr/Fe ratio the strength of the precipitate increases non-monotonously.
- A different Fe-Cr atomic arrangement in 0.6 Cr/Fe precipitate shows a significant change in release stress.
- The length of straight mixed dislocation increases with the strength of precipitate.



# Summary

- Interaction of an edge dislocation with void, He bubble and Cr precipitate were investigated using MD simulation in bcc-Fe.
- Interaction mechanism may differ from one to other by forming an upper jog, a lower jog or no jog formation, respectively.
- With increasing temperature, the release stress decrease for all defects.
- Pressurized He (inert gas) bubbles : the surrounding stress field generated by the bubble has a strong influence on its strength.
- The dislocation loops punched out from the pressurized He bubble interact with the mobile dislocation.
- Cr atomic configuration in Cr precipitates has a strong effect on strengthening.
- For a given size and temperature, a void and a He bubble (even at moderate contents) are stronger obstacle for an edge dislocation than a Cr precipitate.